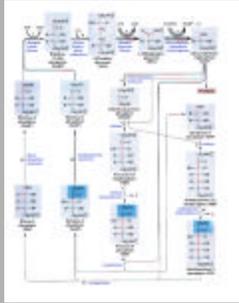
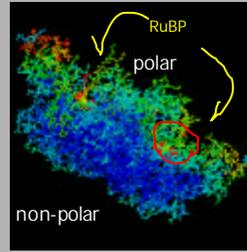


14: Calvin-Benson Cycle



Rubisco



- Photosynthesis: fixation of carbon using metabolic energy derived from photons

- Light reactions: conversion of photons to ATP and NADPH equivalents
- Dark reactions: direct incorporation of CO₂ into carbonyl groups to make new sugar molecules (RuBP → 2GAP)

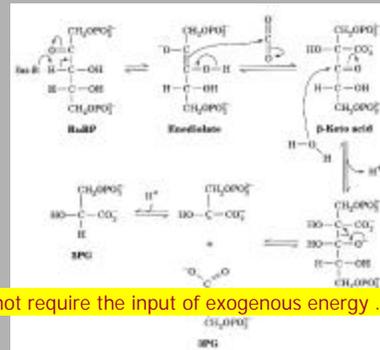
Rubisco reaction does not require the participation of any high-energy reactants

So why the dependence between light and dark reactions?

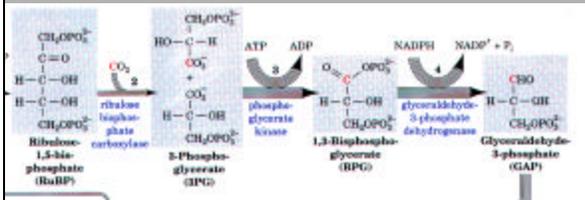
Light Reactions

- PSI & PSII: 4e⁻ excited by 8 photons
- $4\text{Fd}^- + 2\text{NADP}^+ + 2\text{H}^+ \rightarrow 4\text{Fd} + 2\text{NADPH}$
 - FNR requires FAD⁺, which has a E° ~ 0.0V
- Reaction Equivalents:
 - Oxidation of water by the OEC generates 4 H⁺ in the interior for every 4 e⁻ removed to P680⁺.
 - Q-cycle transports 8 H⁺ with those 4 e⁻
- ATP synthase stoichiometry:
 - 3 H⁺ translocated to exterior per ADP → ATP
 - So 12 H⁺ interior will yield 4 ATP
- NADPH has the energy equivalents of 3 ATP
- TOTAL ENERGY YIELD: 10 ATP per 8hm
 - 4 ATP & 2 NADPH per 8hm

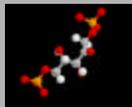
Dark Reactions: CO₂ fixation



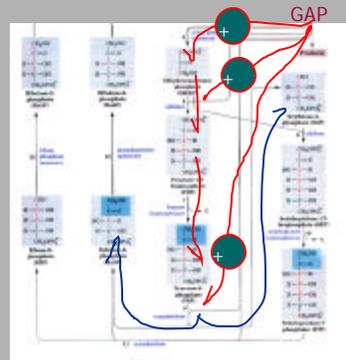
Calvin-Benson: Reaction Series 1



- Cumulative $\Delta G^\circ = -17.1 \text{ kJ mol}^{-1}$

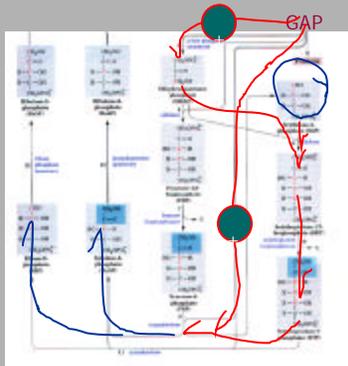


Calvin-Benson: Reaction Series 2



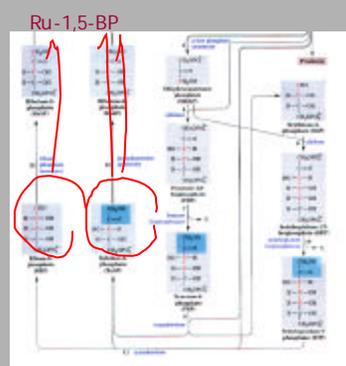
- $\text{C}_3 + \text{C}_3 \rightarrow \text{C}_6$
- $\text{C}_6 + \text{C}_3 \rightarrow \text{C}_9$
- $\text{C}_9 \rightarrow \text{C}_4 + \text{C}_5$
- $\Sigma \Delta G^\circ$
- $-37.2 \text{ kJ mol}^{-1}$

Calvin-Benson: Reaction Series 3



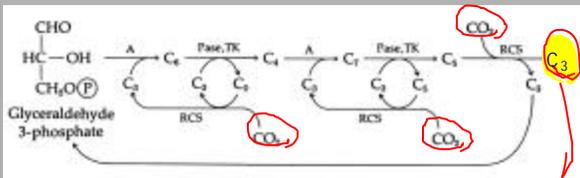
- $\text{C}_3 + \text{C}_4 \rightarrow \text{C}_7$
- $\text{C}_7 + \text{C}_3 \rightarrow \text{C}_{10}$
- $\text{C}_{10} \rightarrow \text{C}_5 + \text{C}_5$
- $\Sigma \Delta G^\circ$
- $-43.1 \text{ kJ mol}^{-1}$

Calvin-Benson: Reaction Series 4



- Regenerate the initial reactant:
- Ru-1,5-BP
- $\Sigma \Delta G^\circ$
- $-18.1 \text{ kJ mol}^{-1}$

Calvin-Benson: Summary



- Functions as a network of reactions rather than a pathway of reactions.

Summary Reaction

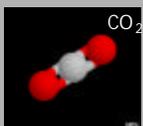
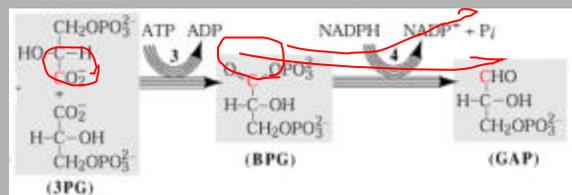
- Start: $3 \text{CO}_2 + 9 \text{ATP} + 6 \text{NADPH}$



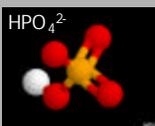
- Finish: $\text{GAP} + 9 \text{ADP} + 6 \text{NADP}^+ + 8 \text{P}_i$



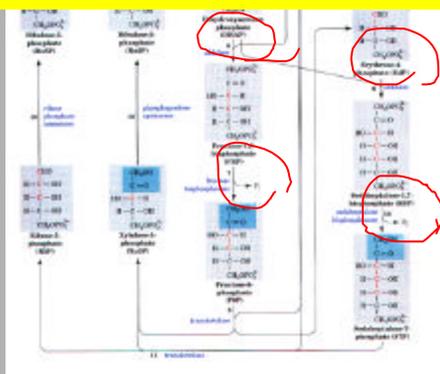
Carbonyl Electrons



Each dephosphorylation results in the loss of electron energy associated with the carbonyl group oxygen



Why the negative ΔG s?



Calvin-Benson: Biosynthesis

Rubisco

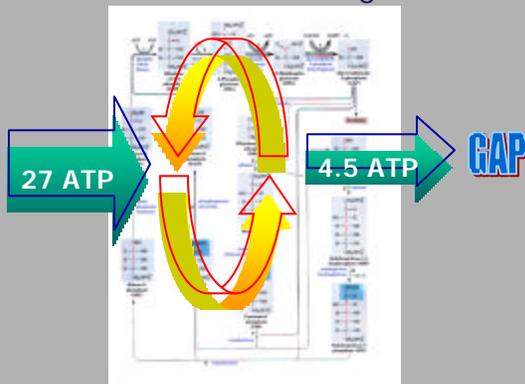


- Calvin Benson Cycle requires a high energy input because it is a biosynthetic (anabolic) pathway
- 90% of the energy input into the cycle is for the conversion of 3PG to GAP
- Yet only 16% of the total energy input will be exported as biosynthetic building blocks (GAP)
- What's wrong with this picture?

Notes:

- Why not fix CO₂ into 3PG and then shunt it into glycolysis? Why invest so much energy into producing GAP?
 - The energy investment in GAP is to regenerate the reaction constituents. Only 16% of the energy input goes into GAP export; the rest serves to reconstruct the RuBP substrate. If 3PG was catabolized in glycolysis, the dark reactions would not be cyclical.

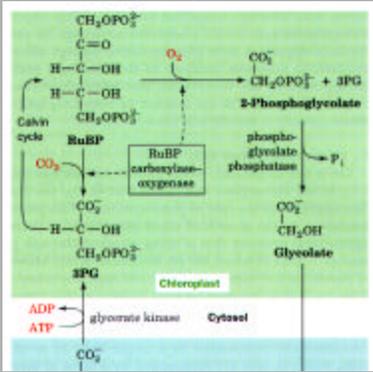
Calvin-Benson: Reagonomics



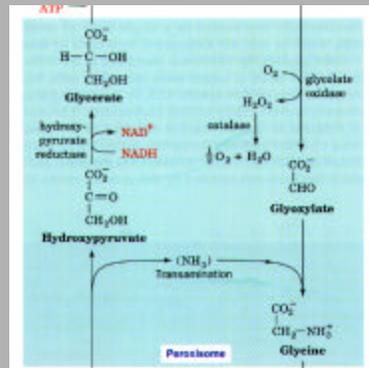
Photorespiration

- The CB Cycle requires a large input of energy
- Rubisco is not 100% at selecting CO₂ from O₂
- The incorporation of O₂ imposes additional energy costs on the cycle
- It's a metabolite recycling cost that requires a decarboxylation event
- For terrestrial organisms, the problem is severe:
 - [CO₂] in air=0.035% [O₂] in air=21%
- For aquatic organisms,
 - the ratio pCO₂:pO₂ is 1/24.

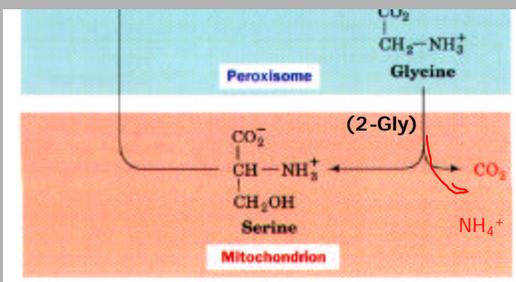
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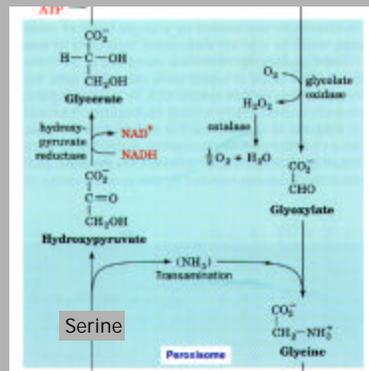
Photorespiration: 2



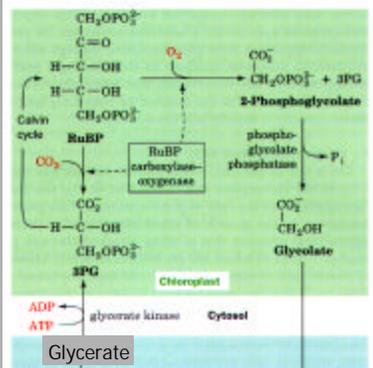
Photorespiration: 3



Photorespiration: 4



Photorespiration:5



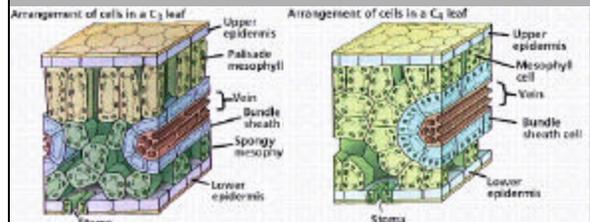
Photorespiration:

- In addition to the low efficiency of energy output by the Calvin-Benson cycle, the less than ideal specificity of Rubisco for CO_2 means that for every O_2 it incorporates, yet an additional 4 ATP have to be spent in recycling the glyoxylate to 3PG.
- PLUS another 12 ATP equivalents are lost during the decarboxylation of a CO_2 that had been previously fixed

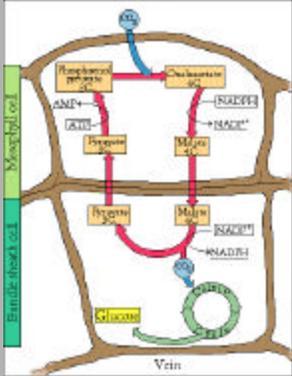
C3 and C4

- BUT, apparently in the plant kingdom, photosynthetic efficiencies using 3-carbon sugars just aren't low enough, so some plants expend even more energy using a 4-carbon intermediate pathway.

C3 and C4 plants

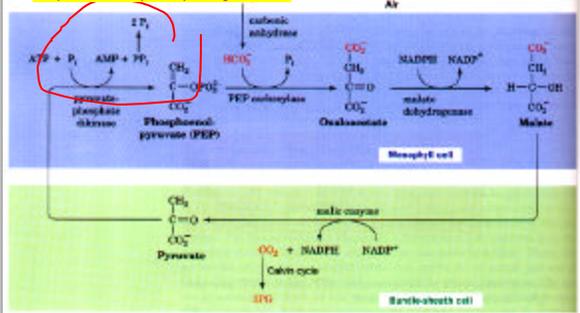


C4 Diagram



C4 Pathway: Hatch-Slack

Equals 2 dephosphorylations

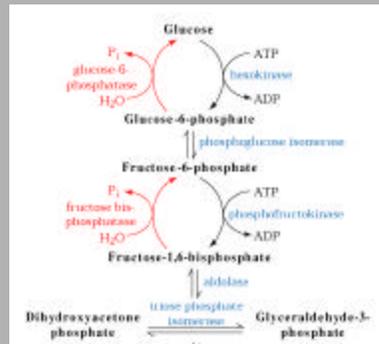


For every CO2 fixed, an additional 2 ATP are required

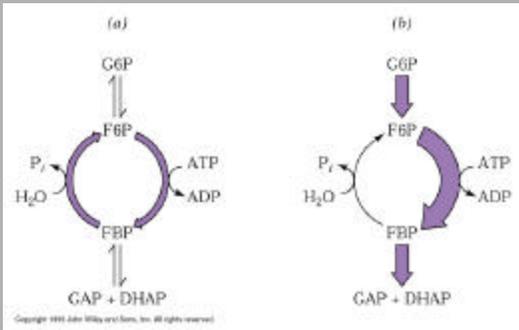
Photosynthetic Energy Balance

- PSI & PSII:
 - TOTAL ENERGY YIELD: 10 ATP per 8h
- It takes 27 ATP equivalents to turn the CB cycle.
- 24 photons to generate 1 GAP for use in biosynthetic activities

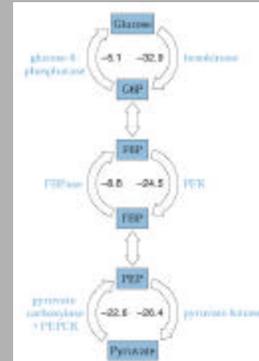
Gluconeogenesis



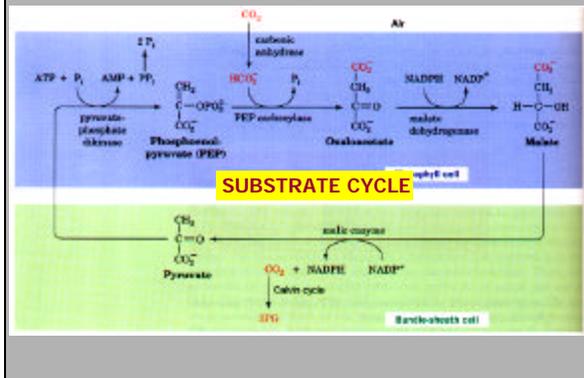
Substrate Cycle



Glycolytic Substrate Cycles

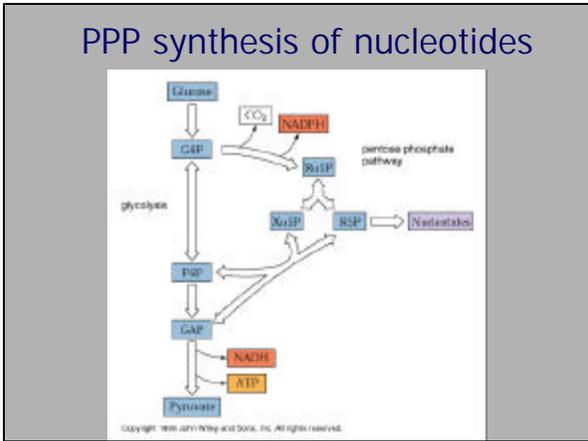
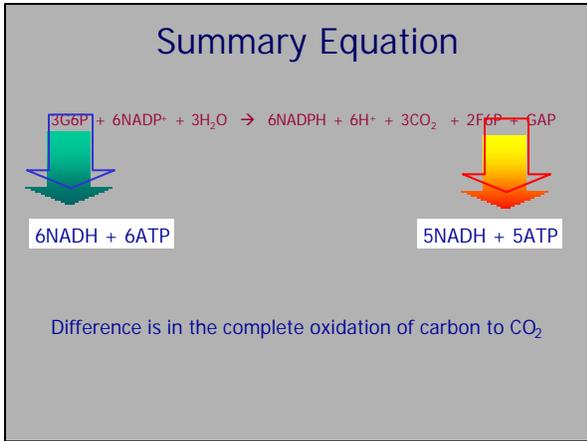
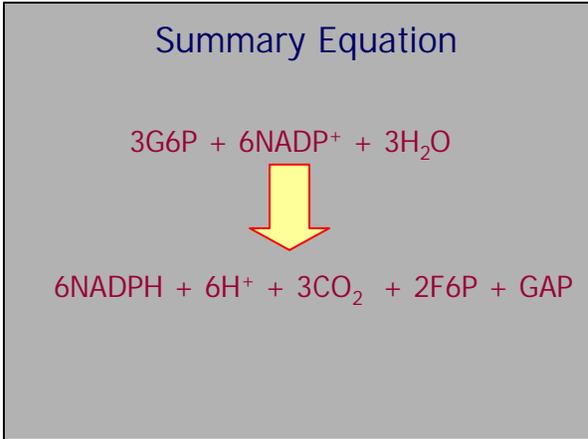
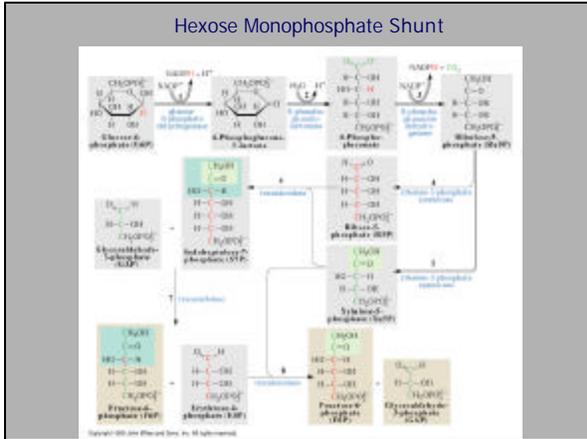


C4 Cycle

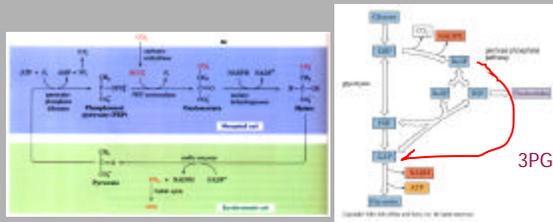


Pentose Phosphate Pathway

- NADPH is used for reductive biosynthesis
 - Fatty acid synthesis
- NADH use primarily for oxidative phosphorylation
 - ATP synthesis
- Intracellular concentrations:
 - $[\text{NAD}^+]/[\text{NADH}] = 1000$; favors metabolite oxidation
 - $[\text{NADP}^+]/[\text{NADPH}] = 0.01$; favors synthetic reduction
- Similar metabolic landscape as in the CB cycle



Early metabolic scheme ??



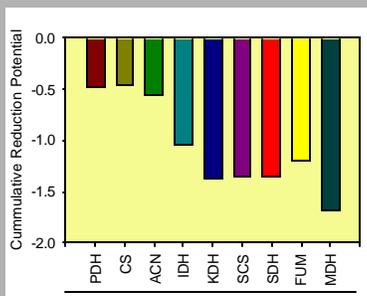
Carbon Cycle Summary

Why is CO₂ so chemically important ?

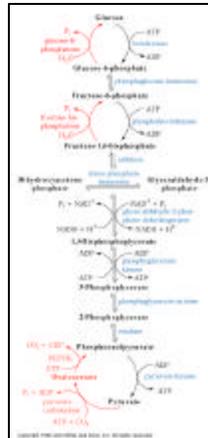


- Summary equation is the same from any starting point in the cycle (which is why it is a cycle).
- The TCA cycle is a series of energy transfer reactions.
- The primary chemical modification is the conversion of $CH_3CH=O + H_2O \rightarrow 2CO_2$
- Essentially this could be called the carbonyl cycle because of the importance of this carbon group in determining the overall energetics of the chemical reactions.

Net change in reduction potential



- Total drop in reduction potential through the TCA cycle is -1.1 V (equivalent to photosystem e⁻ transfers)



Gluconeogenesis

Why can't glycolysis
Be reversed to fix CO₂?

Reductive TCA cycle ???

C=O bond has high e- energy

- In a sense, the forward direction of the TCA cycle is an electron transfer scheme in which the carbonyl group provides the 'electron-excitation' event to influence chemical interactions.
 - Destabilizing force for bond reformation
- The reverse direction, would require a significant level of energy input to overcome the necessary +1.1 V change in reduction potentials.
- In metabolism, C-C bonds can rarely be broken without the direct participation of an adjoining carbonyl group.
- Ex: Glutaraldehyde $\text{CH}_3\text{CH}=\text{O}$; Formaldehyde $\text{H}_2\text{C}=\text{O}$.